



**PROJECT  
ZIA-PHASES**

# **The emergence of ZIA phases**

**M.A. Tunes, H.V. Tin, R.I. Neto, J.A. Valdez,  
S. Fensin, J.K. Baldwin, O. El-Atwani and S.A. Maloy  
Los Alamos National Laboratory**

# Historical background



Jean Discart  
*L'Atelier de Poterie*  
Tangier, Morocco



# Reactive Hot Pressing (RHP) and MAX Phases

392

NATURE

April 24, 1965 VOL. 206

## CERAMICS

### 'Reactive Hot Pressing': a New Ceramic Process

REACTIVE hot pressing is a new process by which powdered materials can be densified and materials of considerable strength can be produced. The term 'reactive hot pressing' is new and has been coined to distinguish this process from other known ceramic processes such as sintering, hot pressing or pressure fabrication. (3,000–15,000 lb./in.<sup>2</sup>) on the 'reactive' solid a considerable interparticle bonding can be achieved in different ceramic materials. Thus, this new process combines the reactivity of a solid during a phase change with pressure to obtain a very strong and dense material. In the present investigation, hydrated silicates, hydroxides, hydrated oxides and carbonates were used. The products after decomposition were oxides and silica in all cases.

A. C. D. CHAKLADER

Department of Metallurgy,  
University of British Columbia,  
Vancouver 8, Canada.

Journal

J. Am. Ceram. Soc., 79 (7), 1955-56, 1996.

J. Am. Ceram. Soc., 79 (7) 1955-56 (1996)

Synthesis and Characterization of a Remarkable Ceramic: Ti<sub>3</sub>SiC<sub>2</sub>

Michel W. Barsoum\* and Tamer El-Raghy\*

Department of Materials Engineering, Drexel University, Philadelphia, Pennsylvania 19104

Polycrystalline bulk samples of Ti<sub>3</sub>SiC<sub>2</sub> were fabricated by reactively hot-pressing Ti, graphite, and SiC powders at 40 MPa and 1600°C for 4 h. This compound has remarkable properties. Its compressive strength, measured at room temperature, was 600 MPa, and dropped to 260 MPa at 1300°C in air. Although the room-temperature failure was brittle, the high-temperature load-displacement curve shows significant plastic behavior. The oxidation is parabolic and at 1000° and 1400°C the parabolic rate constants were, respectively,  $2 \times 10^{-8}$  and  $2 \times 10^{-5}$  kg<sup>2</sup>·m<sup>-4</sup>·s<sup>-1</sup>. The activation energy for oxidation is thus  $\approx 300$  kJ/mol. The room-temperature electrical conductivity is  $4.5 \times 10^6$  Ω<sup>-1</sup>·m<sup>-1</sup>, roughly twice that of pure Ti. The thermal expansion coefficient in the temperature range 25° to 1000°C, the room-temperature thermal conductivity, and the heat capacity are respectively,  $10 \times 10^{-6}$  °C<sup>-1</sup>, 43 W/(m·K), and 588 J/(kg·K). With a hardness of 4 GPa and a Young's modulus of 320 GPa, it is relatively soft, but reasonably stiff. Furthermore, Ti<sub>3</sub>SiC<sub>2</sub> does not appear to be susceptible to thermal shock; quenching from 1400°C into water does not affect the postquench bend strength. As significantly, this compound is as readily machinable as graphite. Scanning electron microscopy of polished and fractured surfaces leaves little doubt as to its layered nature.

Monatshefte Für Chemie, 98(2), 329–337, 1967

Die Kristallstruktur von Ti<sub>3</sub>SiC<sub>2</sub>—ein neuer Komplexcarbidgebiet

Von

W. Jeitschko und H. Nowotny

Aus dem Institut für physikalische Chemie der Universität Wien

Mit 5 Abbildungen

(Eingegangen am 7. Dezember 1966)

der TiSi<sub>2</sub>-reichen Seite angesetzt. Die Ausgangskomponenten waren Pulver von Titanhydrid (Metal Hydrides Inc., Mass., Grade E), Silicium (Péchiney, 99,9%) und Reaktorgraphit. Die Pulvermischungen wurden in abgeschlossener Zelle bei 1600°C und 40 MPa (3000–15000 lb./in.<sup>2</sup>) aufgeschmolzen. Die Titan-Atome besetzen die Punktlagen 2a) und 2b) und die Silicium-Atome die Punktlage 2b) und 4f) ( $z_C = 0,5675$ ) in der Struktur gehört zu den Kristallelementen [Ti<sub>3</sub>SiC<sub>2</sub>].

The crystal structure of Ti<sub>3</sub>SiC<sub>2</sub> was determined by means of single crystal photographs. The hexagonal cell was found with  $a = 3,068$  Å and  $c/a = 5,759$ . The titanium atoms occupy the sites 2a) and 2b) and the silicon atoms 4f) ( $z_C = 0,5675$ ) of the crystal structure type belonging to the space group P6<sub>3</sub>/m. The structure belongs to the layered type having octahedral groups [Ti<sub>3</sub>SiC<sub>2</sub>].

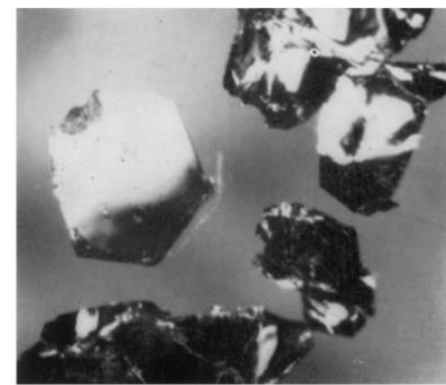
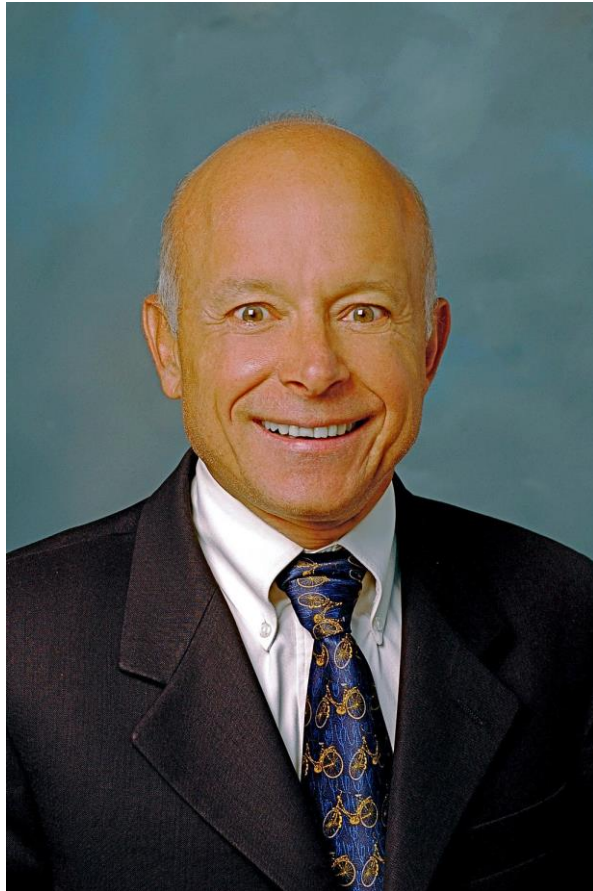


Abb. 1. Unverformter Kristall (hexagonale Symmetrie) und verformte Kristalle von Ti<sub>3</sub>SiC<sub>2</sub>, 50-fach

# Professor Barsoum's MAX Phases Empire in 2021



Source: Wikipedia.org



**M**<sub>n+1</sub>**A****X**<sub>n</sub>

1A												8A					
1 H 1.00794	2A												2 He 4.002602				
3 Li 6.941	4 Be 9.012182											5 B 10.811	6 C 12.0107	7 N 14.0067	8 O 15.9994	9 F 18.9984032	10 Ne 20.1797
11 Na 22.989769	12 Mg 24.3050											13 Al 26.9815386	14 Si 28.0855	15 P 30.973762	16 S 32.065	17 Cl 35.453	18 Ar 39.948
19 K 39.0983	20 Ca 40.078	21 Sc 44.955912	22 Ti 47.867	23 V 50.9415	24 Cr 51.9961	25 Mn 54.938045	26 Fe 55.845	27 Co 58.933195	28 Ni 58.6934	29 Cu 63.546	30 Zn 65.38	31 Ga 69.723	32 Ge 72.64	33 As 74.92160	34 Se 78.96	35 Br 79.904	36 Kr 83.798
37 Rb 85.4678	38 Sr 87.62	39 Y 88.90585	40 Zr 91.224	41 Nb 92.90638	42 Mo 95.96	43 Tc [98]	44 Ru 101.07	45 Rh 102.90550	46 Pd 106.42	47 Ag 107.8682	48 Cd 112.411	49 In 114.818	50 Sn 118.710	51 Sb 121.760	52 Te 127.60	53 I 126.90447	54 Xe 131.293
55 Cs 132.9054519	56 Ba 137.327	57-71 Lanthanides	72 Hf 178.49	73 Ta 180.94788	74 W 183.84	75 Re 186.207	76 Os 190.23	77 Ir 192.217	78 Pt 195.084	79 Au 196.966569	80 Hg 200.59	81 Tl 204.3833	82 Pb 207.2	83 Bi 208.98040	84 Po [209]	85 At [210]	86 Rn [222]
87 Fr 223	88 Ra [226]	89-103 Actinides	104 Rf [267]	105 Db [268]	106 Sg [271]	107 Bh [272]	108 Hs [270]	109 Mt [276]	110 Ds [281]	111 Rg [280]	112 Uub [285]	113 Uut [284]	114 Uuq [289]	115 Uup [288]	116 Uuh [293]	117 Uus [294]	118 Uuo [294]
Lanthanides			57 La 138.90547	58 Ce 140.116	59 Pr 140.90765	60 Nd 144.242	61 Pm [145]	62 Sm 150.36	63 Eu 151.964	64 Gd 157.25	65 Tb 158.92535	66 Dy 162.500	67 Ho 164.93032	68 Er 167.259	69 Tm 168.93421	70 Yb 173.054	71 Lu 174.9668
Actinides			89 Ac [227]	90 Th 232.03806	91 Pa 231.03588	92 U 238.02891	93 Np [237]	94 Pu [244]	95 Am [243]	96 Cm [247]	97 Bk [247]	98 Cf [251]	99 Es [252]	100 Fm [257]	101 Md [258]	102 No [259]	103 Lr [262]

Goossens, N., Tunca, B., Lapauw, T., Lambrinou, K., & Vleugels, J. (2021) MAX phases, structure, processing, and properties. *Encyclopedia of Materials: Technical Ceramics and Glasses*, 2-3, 182-199



TMS 2022 150<sup>th</sup> ANNUAL MEETING & EXHIBITION  
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# We are interested in materials for application in extreme environments: high-temperature and irradiation. And MAX Phases...

Acta Materialia 169 (2019) 237–247



Contents lists available at ScienceDirect

Acta Materialia

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Full length article

## A Transmission Electron Microscopy study of the neutron-irradiation response of Ti-based MAX phases at high temperatures

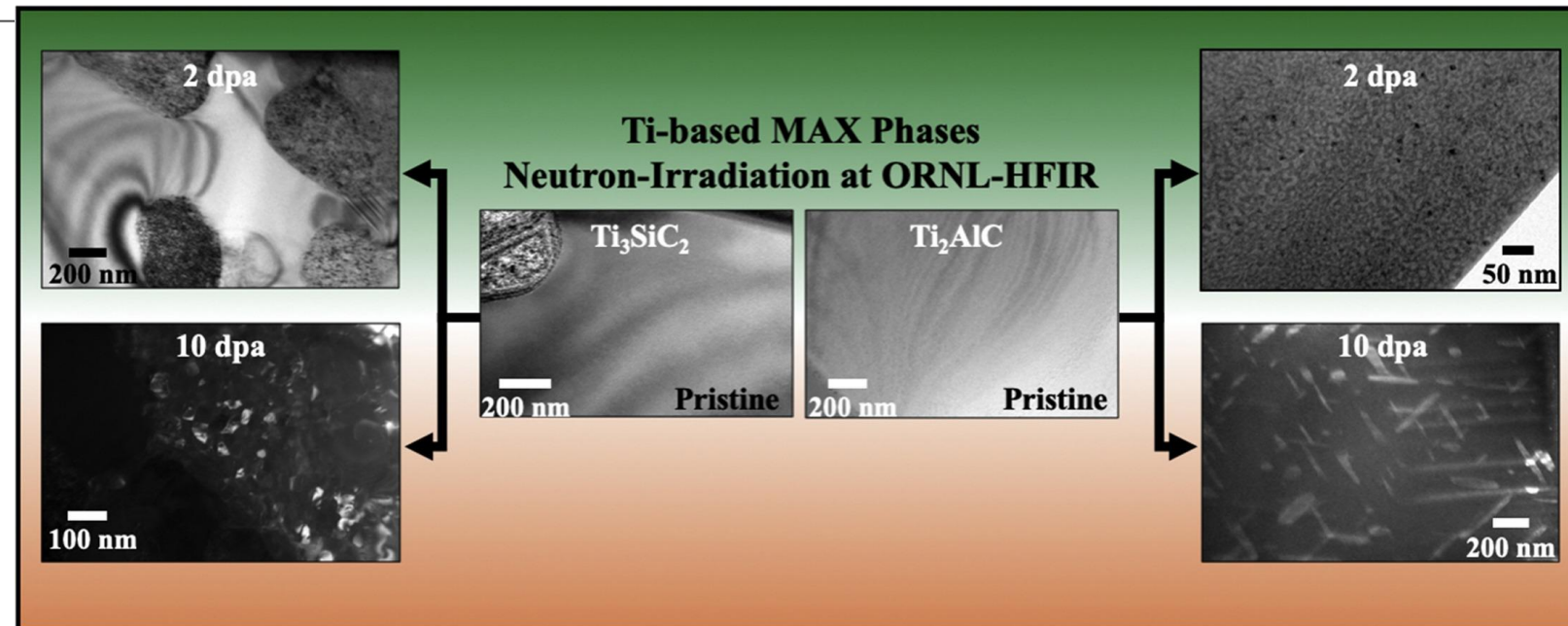


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<sup>b</sup> School of Mechanical, Aerospace and Civil Engineering, The University of Manchester, Sackville Street, M1 3NJ, Manchester, United Kingdom

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# Project's Outline

- **Work-to-date and main objectives.**
- **The ZIA phases project.**
  - **Synthesis and characterization: SEM, EDX, and TEM**
- **Potential of ZIA phases in extreme environments.**

# Work-to-date



# Can we extend the MAX phase concept beyond carbides and nitrides? Yes, intermetallic silicides!

THE CRYSTAL STRUCTURES OF  $Mn_3Ni_2Si$ ,  $V_3Ni_2Si$ ,  
 $Nb_3Ni_2Si$  AND RELATED Cr AND Ta COMPOUNDS

E. I. Gladyshevskii, Yu. B. Kuz'ma, and P. I. Kripyakevich

I. Franko L'vovsk State University

Translated from Zhurnal Strukturnoi Khimii, Vol. 4, No. 3,

pp. 372-379, May-June, 1963

Original article submitted February 6, 1962

The existence was established of ternary  $R_3Ni_2Si$  compounds ( $R = Mn, V, Nb$ ), of the  $\eta$ -phase structure type (space group  $Fd\bar{3}m - O_h^7$ ,  $Z = 16$ ) with completely ordered distributions of all component atoms. Compounds of the  $\eta$ -phase type, with similar but undetermined compositions, are found in the Cr-Ni-Si and Ta-Ni-Si systems.

together Mn (electrolytic), Ni (99.9%) and Si (99.5%) in a porcelain crucible, using a high-frequency furnace (hydrogen atmosphere). As the alloying was not accompanied by combustion, a chemical analysis was not carried out.

Three specimens were used in the x-ray analysis: 1) a cast specimen (no heat treatment); 2) a specimen annealed at  $1000^\circ$  for 24 hr; 3) a specimen annealed at  $800^\circ$  for 120 hr. Annealing was carried out in evacuated quartz am-

The  $\eta$ -phase in the Nb-Ni-Si system was also obtained in an almost pure state (see Fig. 2).



Fig. 2. Microstructure of the alloy  $Nb_3Ni_2Si$  (composition of materials used). Magnification 200.



# Re-assessing the Nb-Si-Ni system: 2019.

## Can we stabilize the Nb<sub>3</sub>SiNi<sub>2</sub> phase?

Scripta Materialia 164 (2019) 96–100



ELSEVIER

Contents lists available at ScienceDirect

Scripta Materialia

journal homepage: [www.elsevier.com/locate/scriptamat](http://www.elsevier.com/locate/scriptamat)



Experimental investigation of phase equilibria in the Nb–Ni–Si refractory alloy system at 1073 K

Vinícius O. dos Santos<sup>a, b, \*</sup>, Matheus A. Tunes<sup>c</sup>, Luiz T.F. Eleno<sup>d</sup>, Cláudio G. Schön<sup>a</sup>, Klaus W. Richter<sup>b</sup>

<sup>a</sup>Escola Politécnica da Universidade de São Paulo, Department of Metallurgical and Materials Engineering, Av. Prof. Mello Moraes 2463, - São Paulo CEP 05508-030, SP, Brazil

<sup>b</sup>University of Vienna, Faculty of Chemistry, Department of Inorganic Chemistry - Functional Materials, Althanstrasse 14, Vienna 1090, Austria

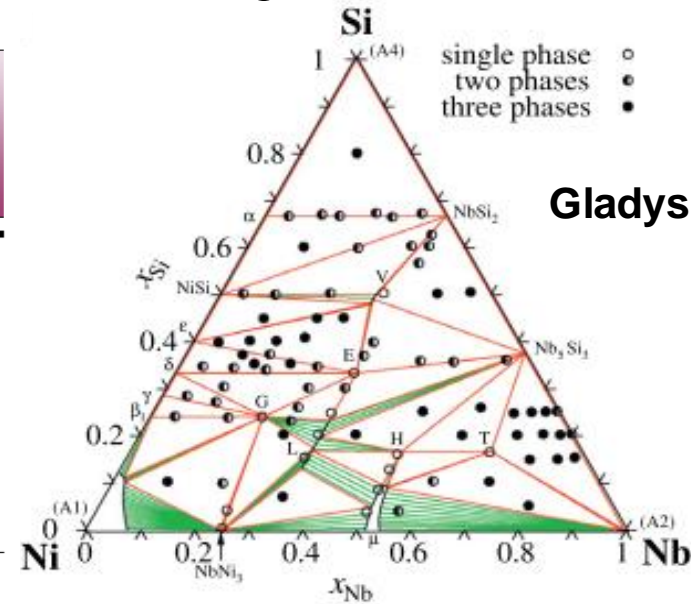
<sup>c</sup>School of Computing and Engineering, University of Huddersfield, United Kingdom

<sup>d</sup>Escola de Engenharia de Lorena da Universidade de São Paulo, Department of Materials Engineering, Brazil

### ABSTRACT

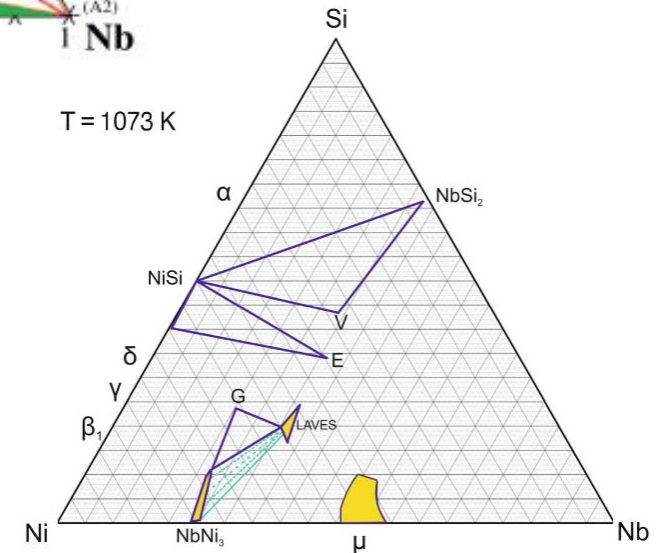
Twelve different compositions of the ternary Nb–Ni–Si system were synthesized and heat-treated at 1073 K for up to 1000 h. The material. Twelve samples were weighed in an analytical balance (resolution ± 0.0005 g) and then arc-melted using tungsten electrode under argon atmosphere. The first series of compositions, V01, latter, due to the di

800°C/120 hours: Gladyshevskii et al. 1962  
800°C/1000 hours: us, 2019.



Gladyshevskii et al. 1962

Corrected phase diagram at 1073 K:  
No H-phase.



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# Re-assessing the Nb-Si-Ni system: 2020.

## We are looking for the H-phase again!

Journal of Alloys and Compounds 842 (2020) 155373



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Journal of Alloys and Compounds

journal homepage: <http://www.elsevier.com/locate/jalcom>



Sample	Sample Composition (at%)			Phase	Phase Composition (at%)			Space Group	Lattice Parameters (Å)		
	Nb	Ni	Si		Nb	Ni	Si		a	B	c
V06T-2	45.56	34.57	19.88	Laves	34.06	44.48	21.44	P63/mmc	4.82		7.79
				T	64.44	13.7	21.87	P4/mcc	6.18		5.03
				H	49.69	34.96	15.35	Fd-3mS	11.17		

Experimental investigation of phase equilibria in the Nb–Ni–Si refractory alloy system at 1323 K

Vinícius O. dos Santos <sup>a, b</sup>, Luiz T.F. Eleno <sup>c</sup>, Cláudio G. Schön <sup>a, \*</sup>, Klaus W. Richter <sup>b</sup>

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<sup>b</sup> University of Vienna, Faculty of Chemistry, Department of Inorganic Chemistry - Functional Materials, Althanstrasse 14, 1090, Vienna, Austria

<sup>c</sup> Escola de Engenharia de Lorena da Universidade de São Paulo, Department of Materials Engineering, Brazil

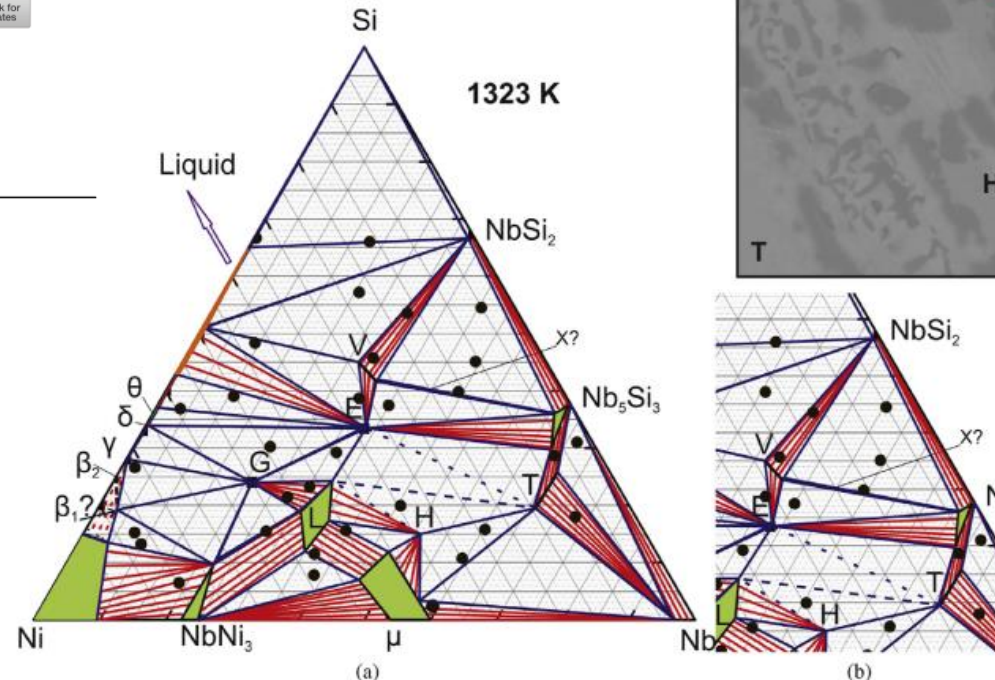


The samples were first arc-melted under Ar using Zr as getter material and then encapsulated in quartz tubes under low pressure of argon and homogenized at 1323 K. The samples were annealed for 336 h in all series. After the prescribed times the capsules were

800°C/120 h: Gladyshevskii et al. 1962

800°C/1000 h: us, 2019.

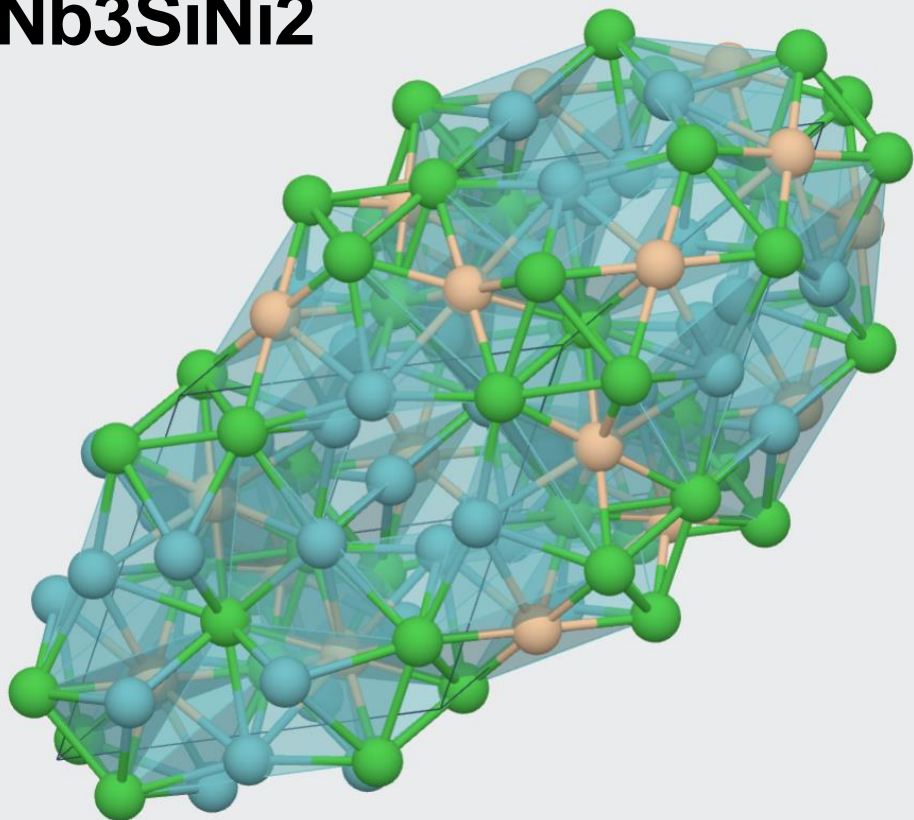
1050°C/336 h: us, 2020.





H-phase appears, but never in its pure form

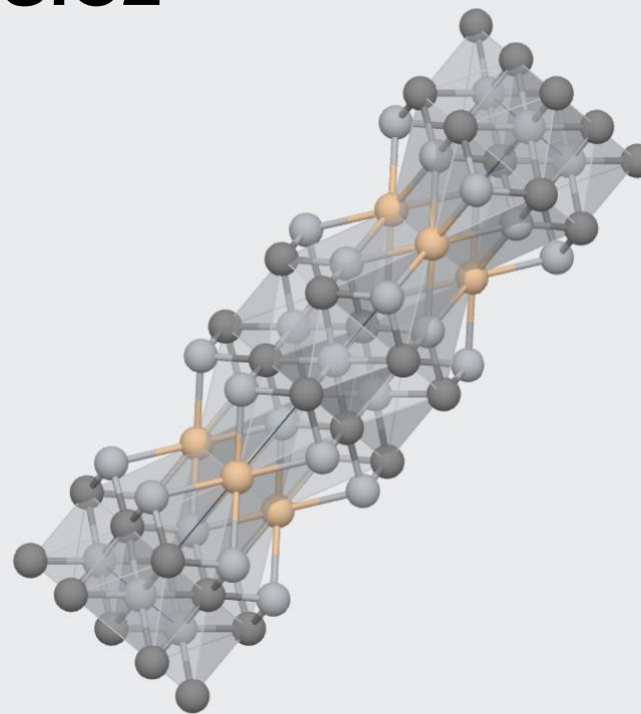




# Nb<sub>3</sub>SiNi<sub>2</sub>



Nb Ni Si  Atoms Unit Cell  Bonds Polyhedra  

# Ti<sub>3</sub>SiC<sub>2</sub>



C Si Ti  Atoms Unit Cell  Bonds Polyhedra  

# The ZIA phases project

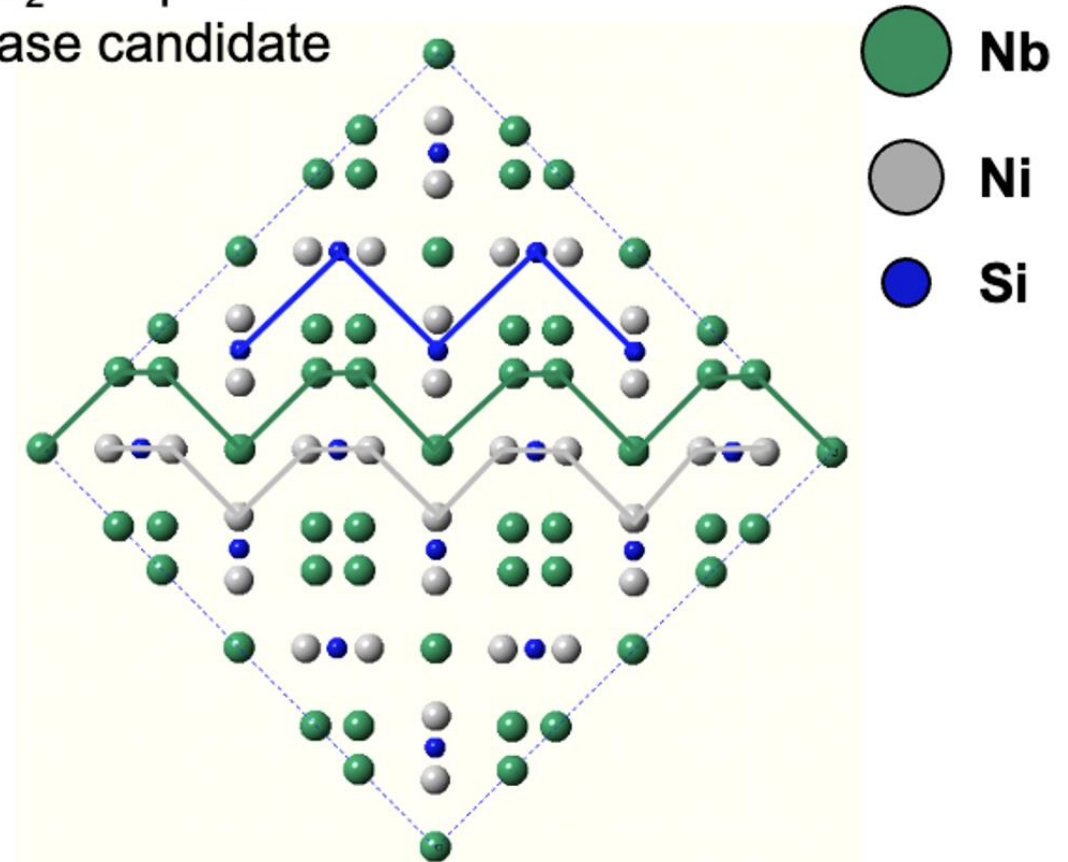


# The ZIA phases project

## Zig-zag Intermetallic Advanced phases

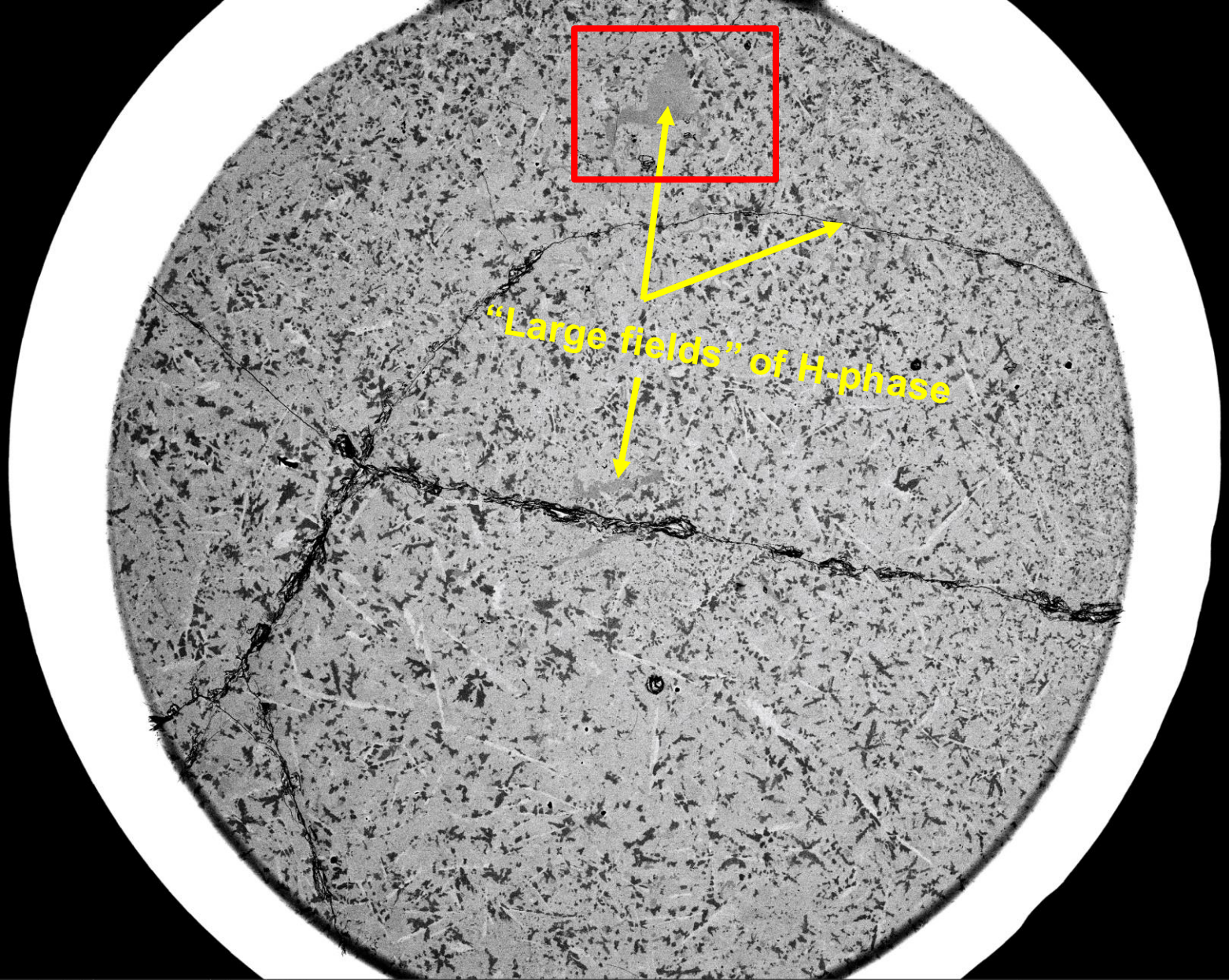
- It is a challenge to synthesize and stabilize the H-phase.
- We brought this problem to LANL.
- We want to understand if:
  - $R_3SiNi_2$  compounds with  $R = Nb, Mo, Ta, Hf, W$  ... represent a novel class of materials.
  - If this novel class can be applied in extreme environments.

$Nb_3SiNi_2$  – H phase  
ZIA phase candidate



# Using arc melting, we produced a sample with the $\text{Nb}_3\text{SiNi}_2$ stoichiometry

First objective:  
Find the H-phase  
( $\text{Nb}_3\text{SiNi}_2$ )



threshold 0 eV use case Standard det CBS HV 20.00 kV WD 9.9 mm curr 1.6 nA mag 75 x dwell 10.00  $\mu\text{s}$

1 mm



# H-phase has been found in the as-cast sample

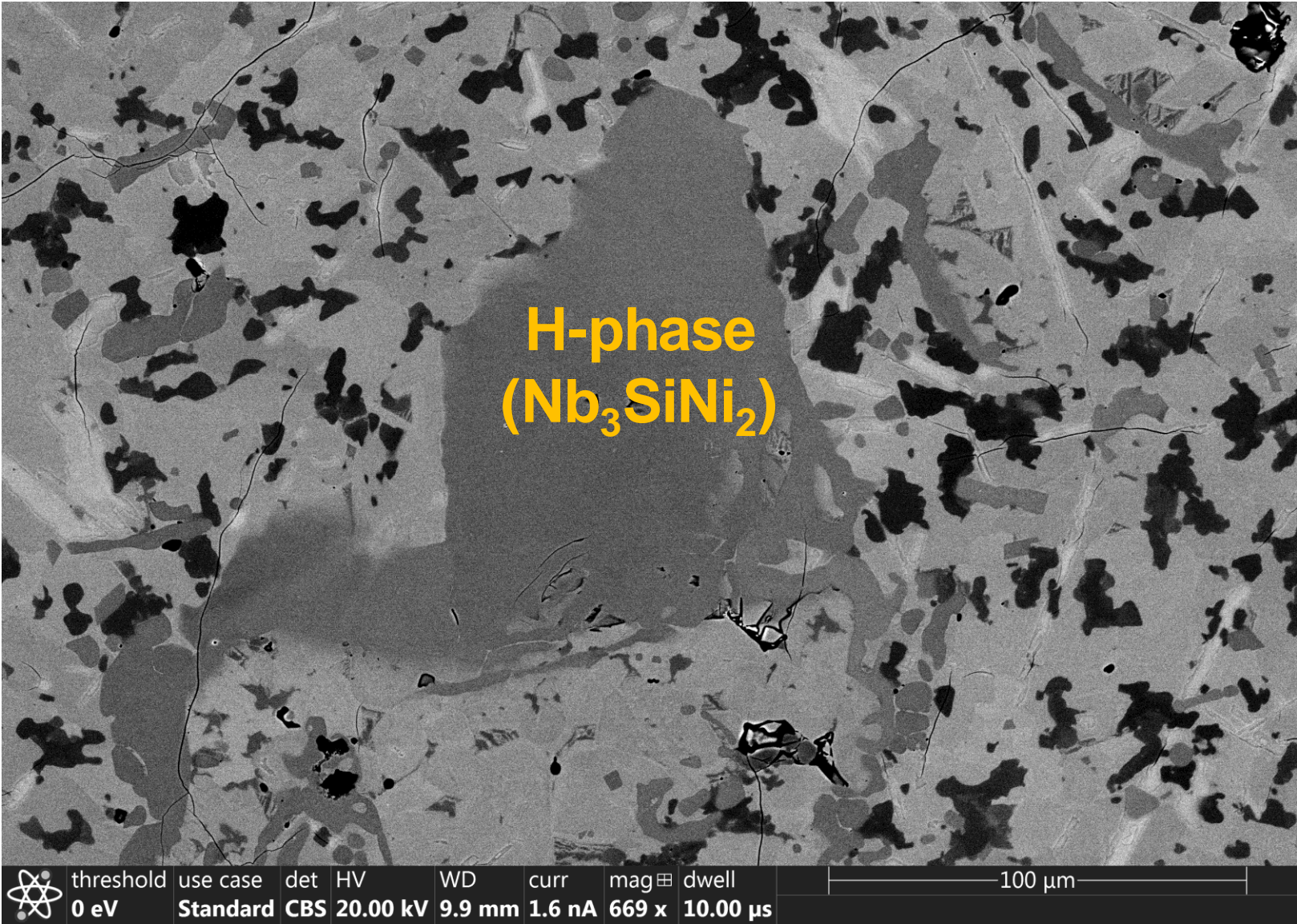
### Expected Compositions

Nb <sub>3</sub> SiNi <sub>2</sub>		H Phase	
	Stoichiometry	at. %	Ratios
Nb	3	50%	3
Si	1	17%	1.5
Ni	2	33%	

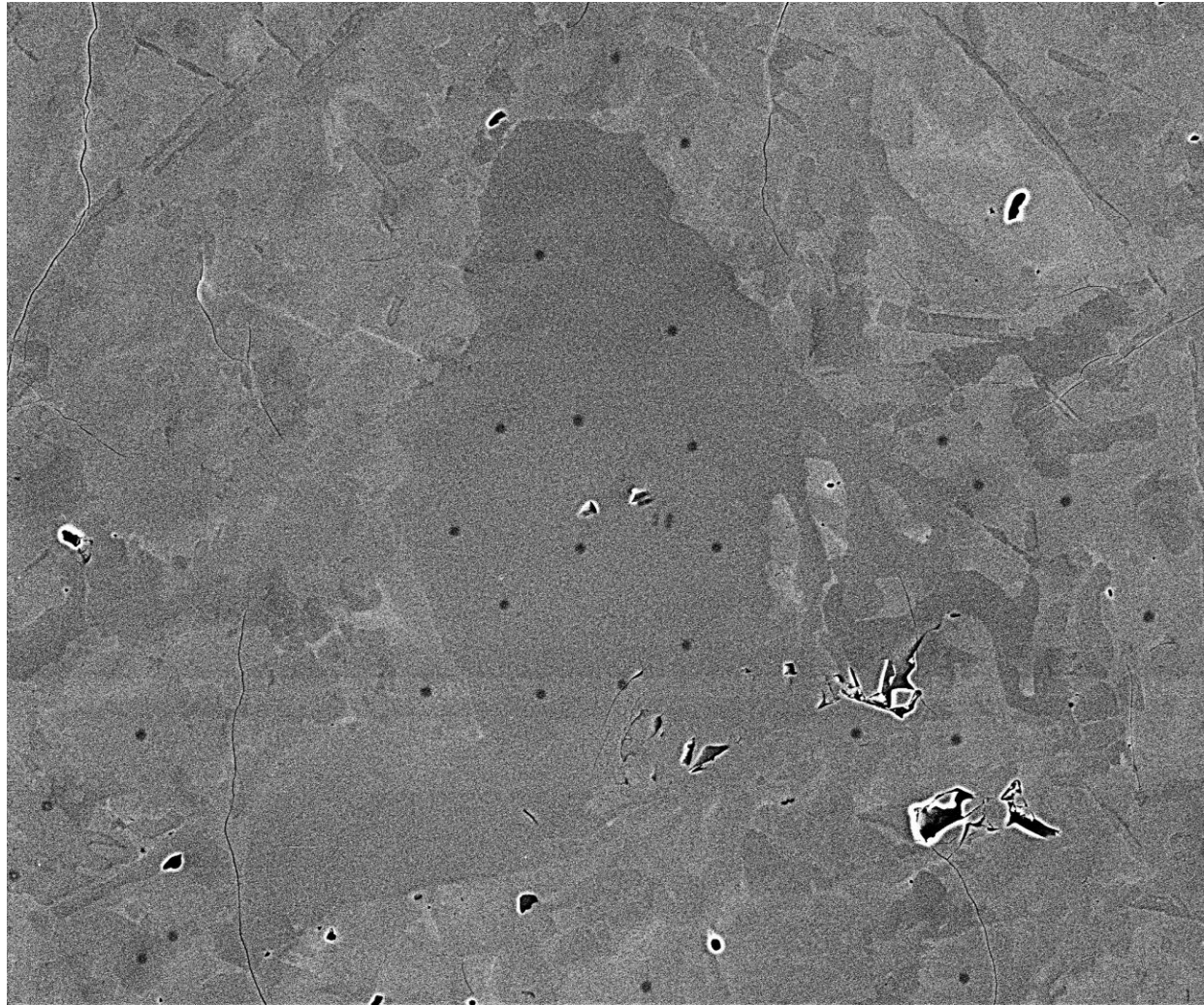
### Measured Compositions

Element	Weight %	MDL	Atomic %	Net Int.	Error %
Si K	8.26	0.04	20.08	4348.72	6.56
Ni K	29.16	0.13	33.92	5116.37	2.68
Nb L	62.59	0.09	46.00	15333.20	5.19

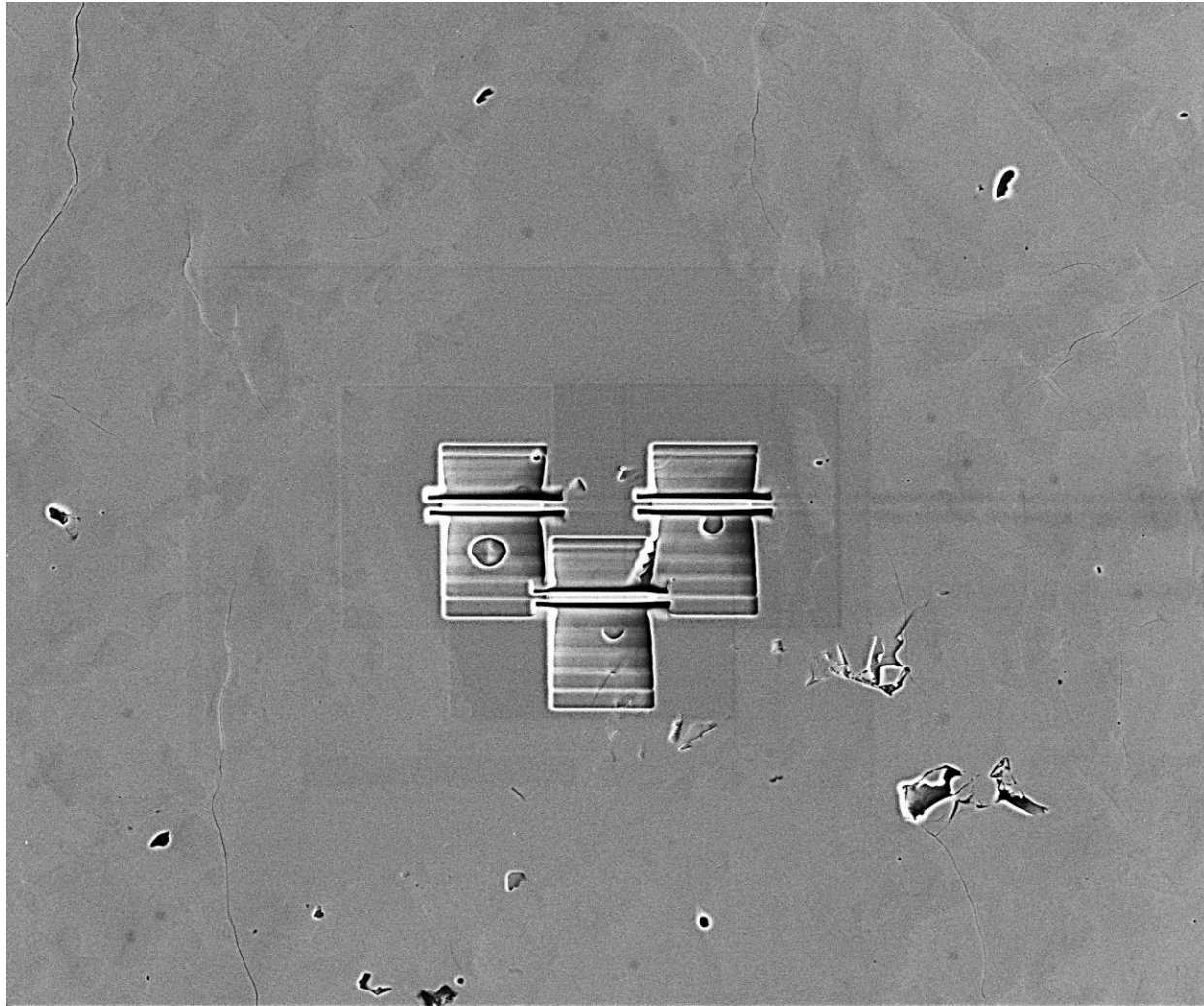
- Si is overestimated by EDX.
- Equilibrium not attained.
- Compositions may also be slightly off.



# We decided to do TEM on the H-phase field (as-cast)

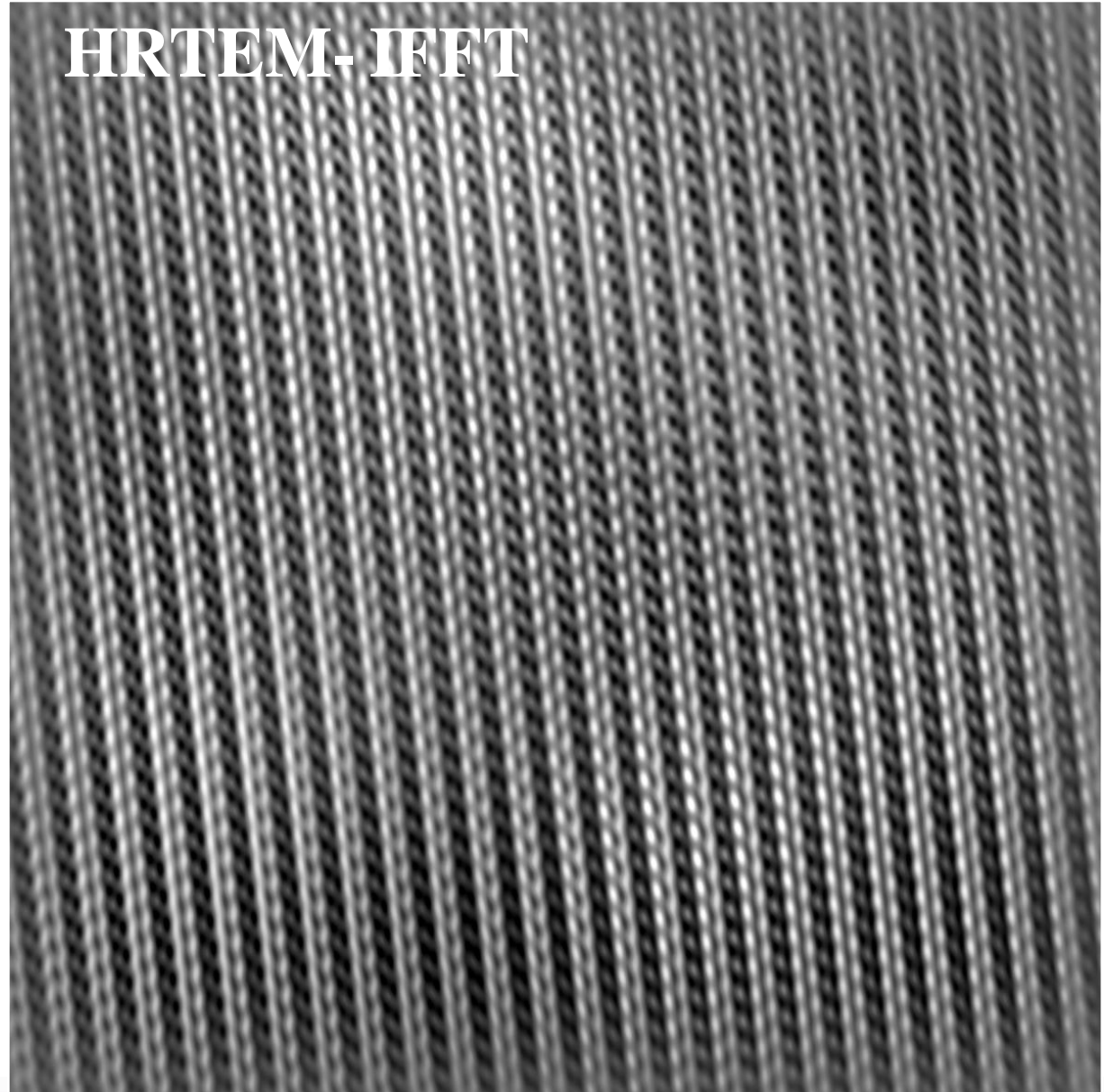
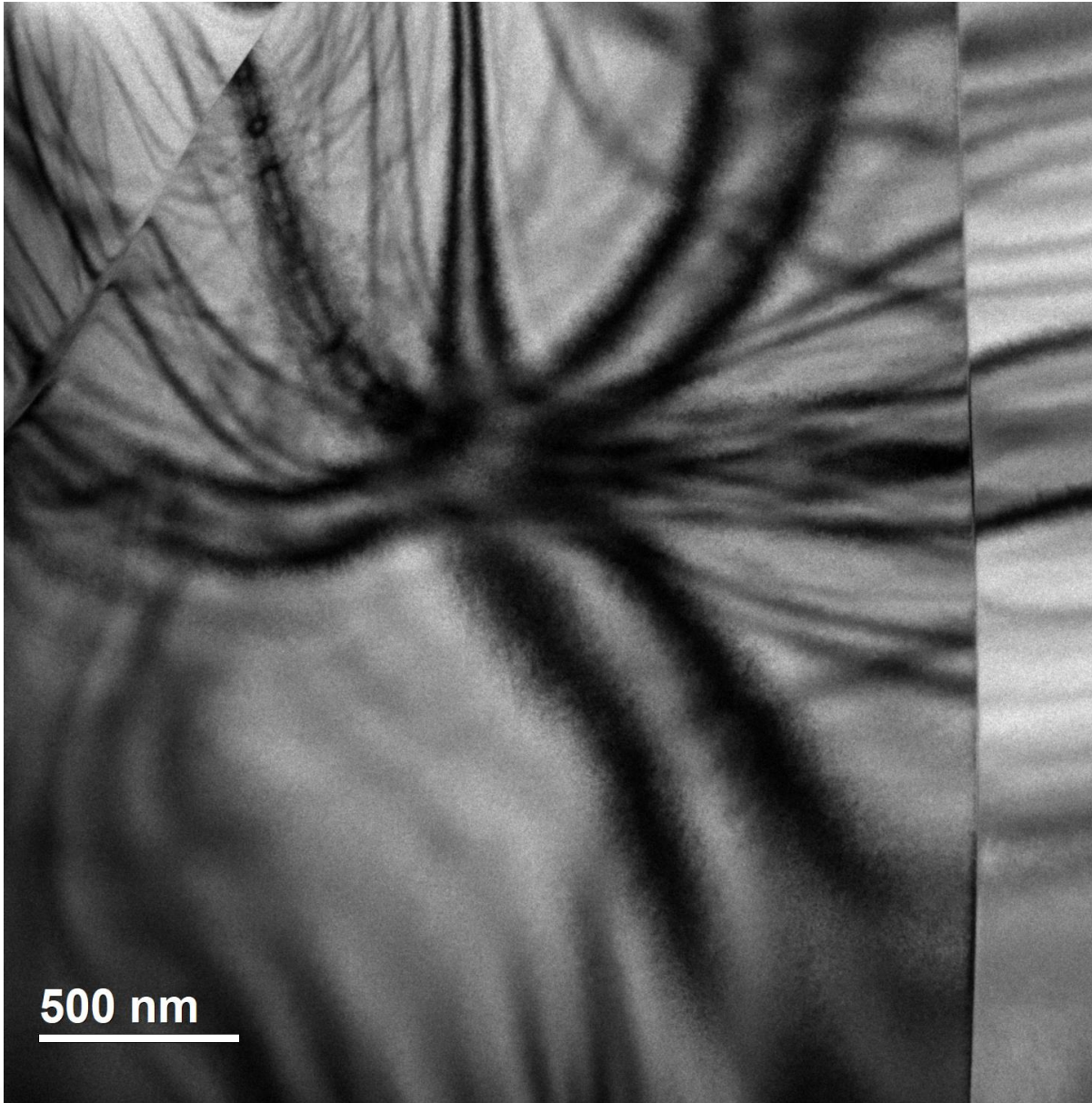


	HV	6/8/2021	mag □	WD	det	50 μm
	20.00 kV	8:27:49 AM	1 200 x	4.6 mm	ETD	Nb <sub>3</sub> NiSi <sub>2</sub>



	HV	6/8/2021	mag □	WD	det	50 μm
	20.00 kV	10:22:20 AM	1 200 x	4.1 mm	ETD	Nb <sub>3</sub> NiSi <sub>2</sub>

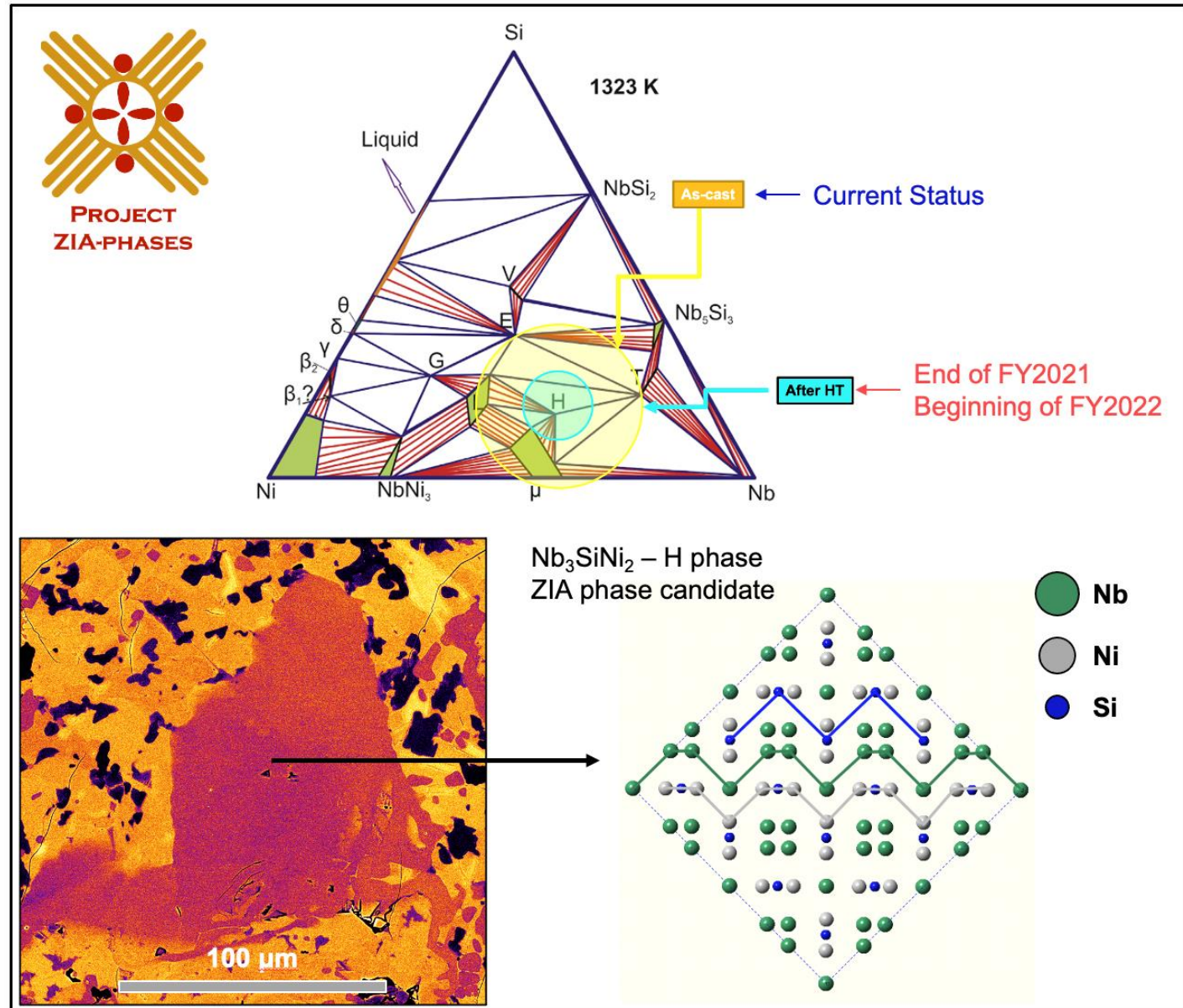
# A distinct nanolayered structure has been observed!





# Summary of the sample as-cast

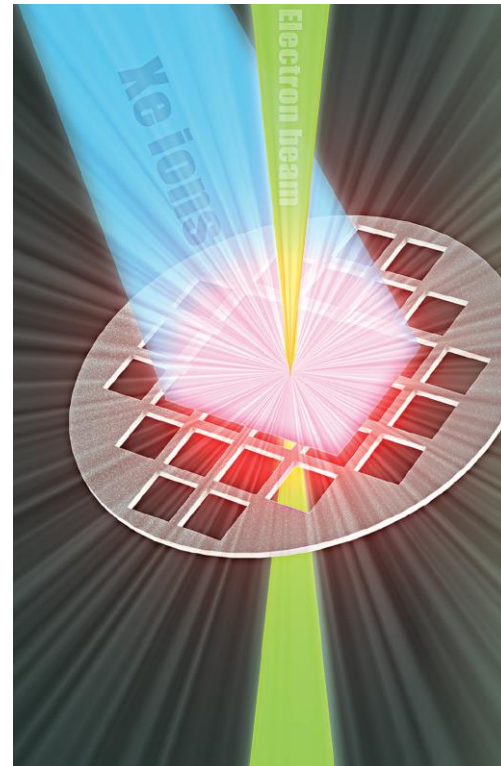
1. The Nb—Si—Ni system with 312 stoichiometry is not under equilibrium.
2. H-phase field has been observed at RT (for the first time) using arc melting.
3. H-phase is indeed a nanolayered superstructure.
4. Next steps:
  - New HT routes.
  - Will the system converge to the H-phase pure?
  - Investigating different synthesis methodologies?



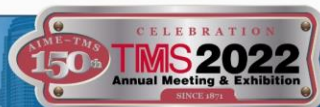
# Potential application of ZIA phases



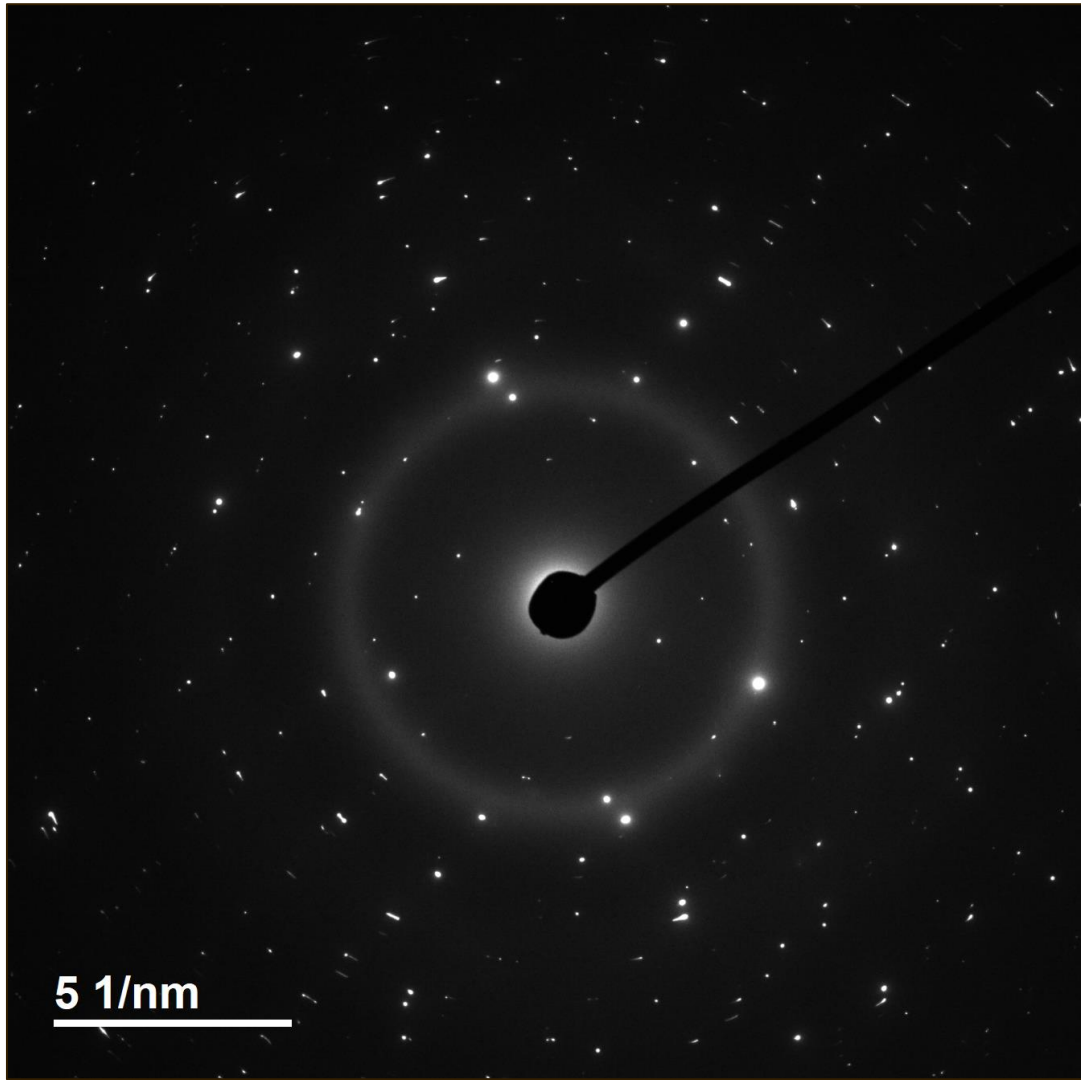
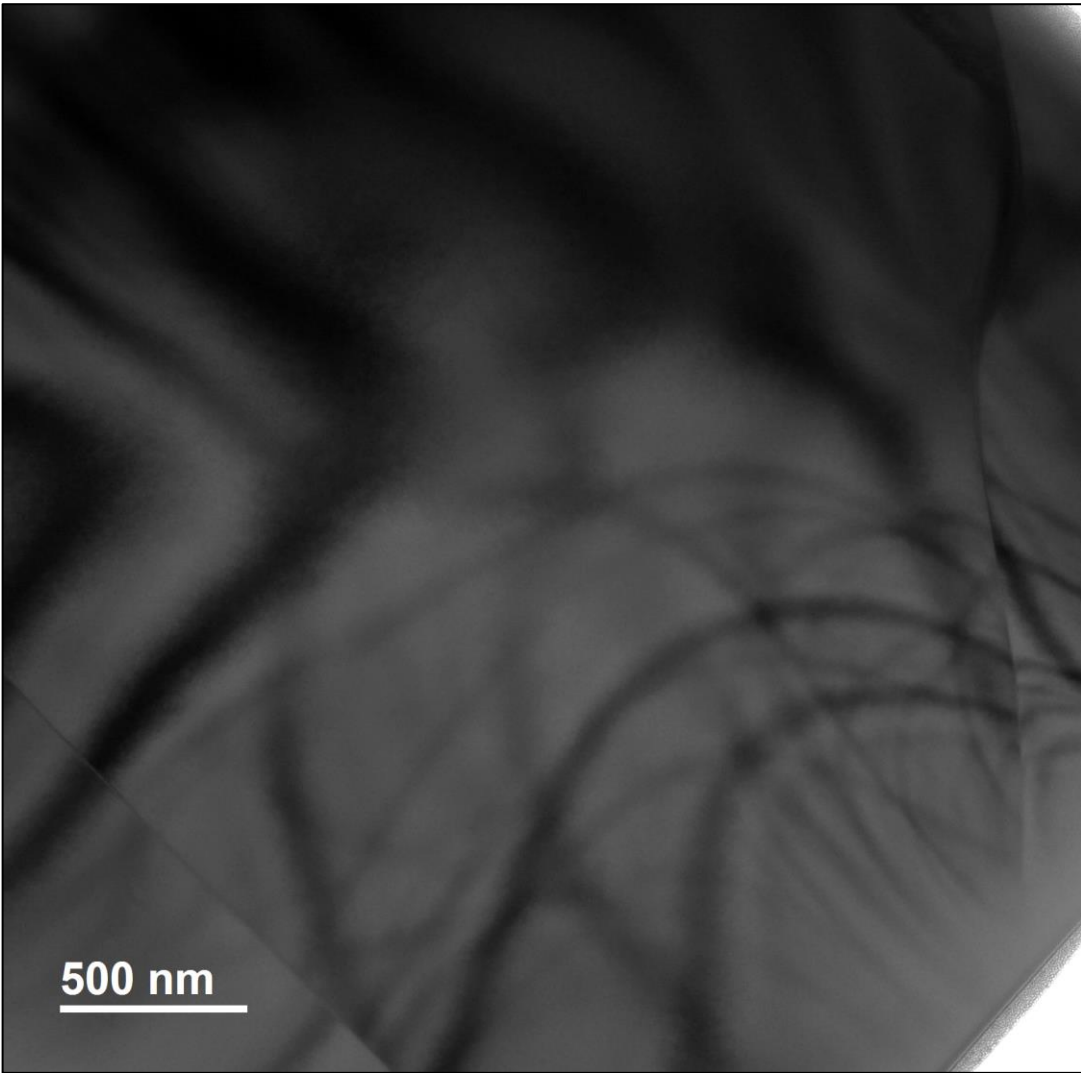
# In situ TEM ion irradiation at the Argonne National Laboratory (IVEM Facility)



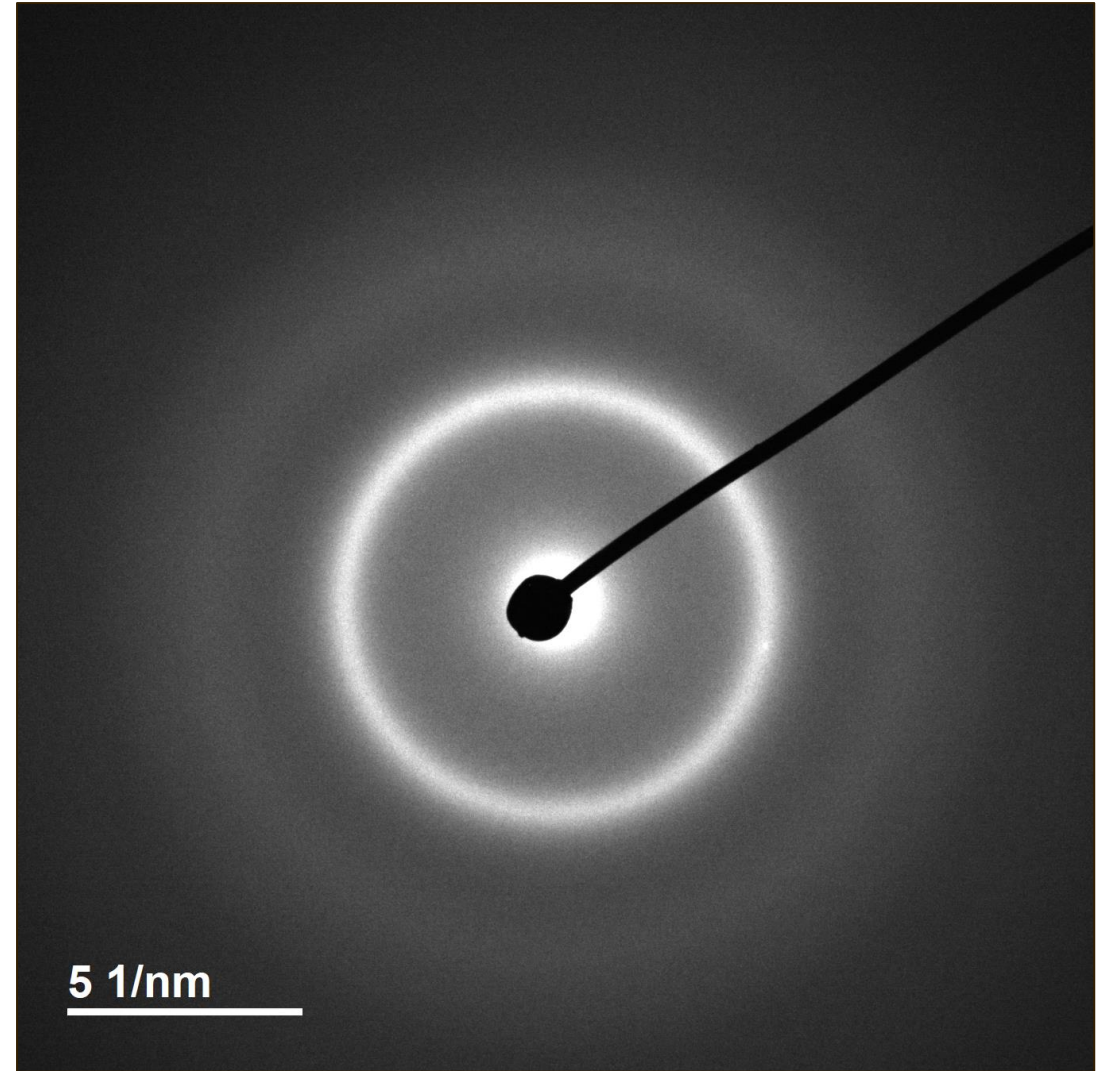
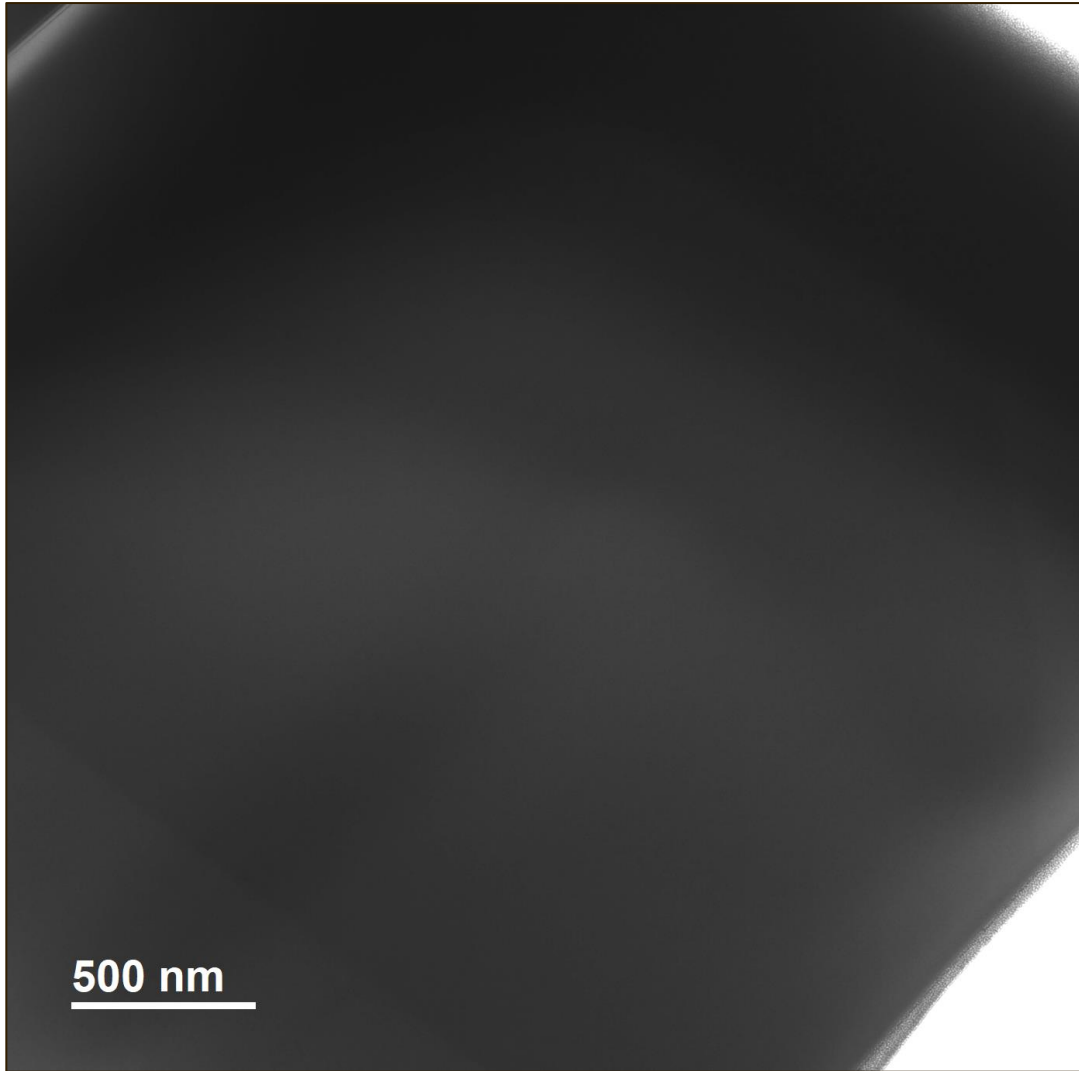
Work in  
partnership with  
Dr. H.V. Tin (LANL)



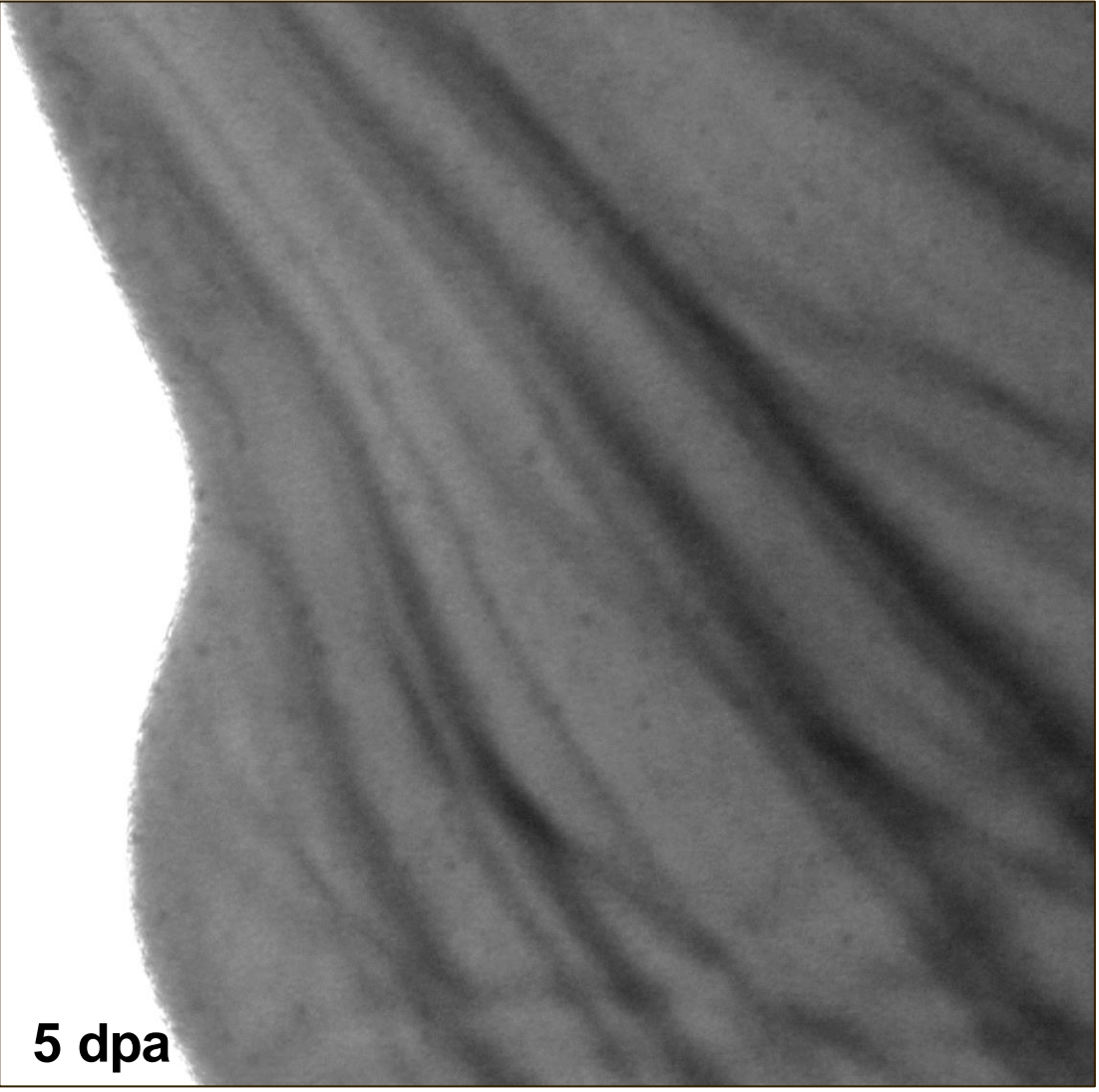
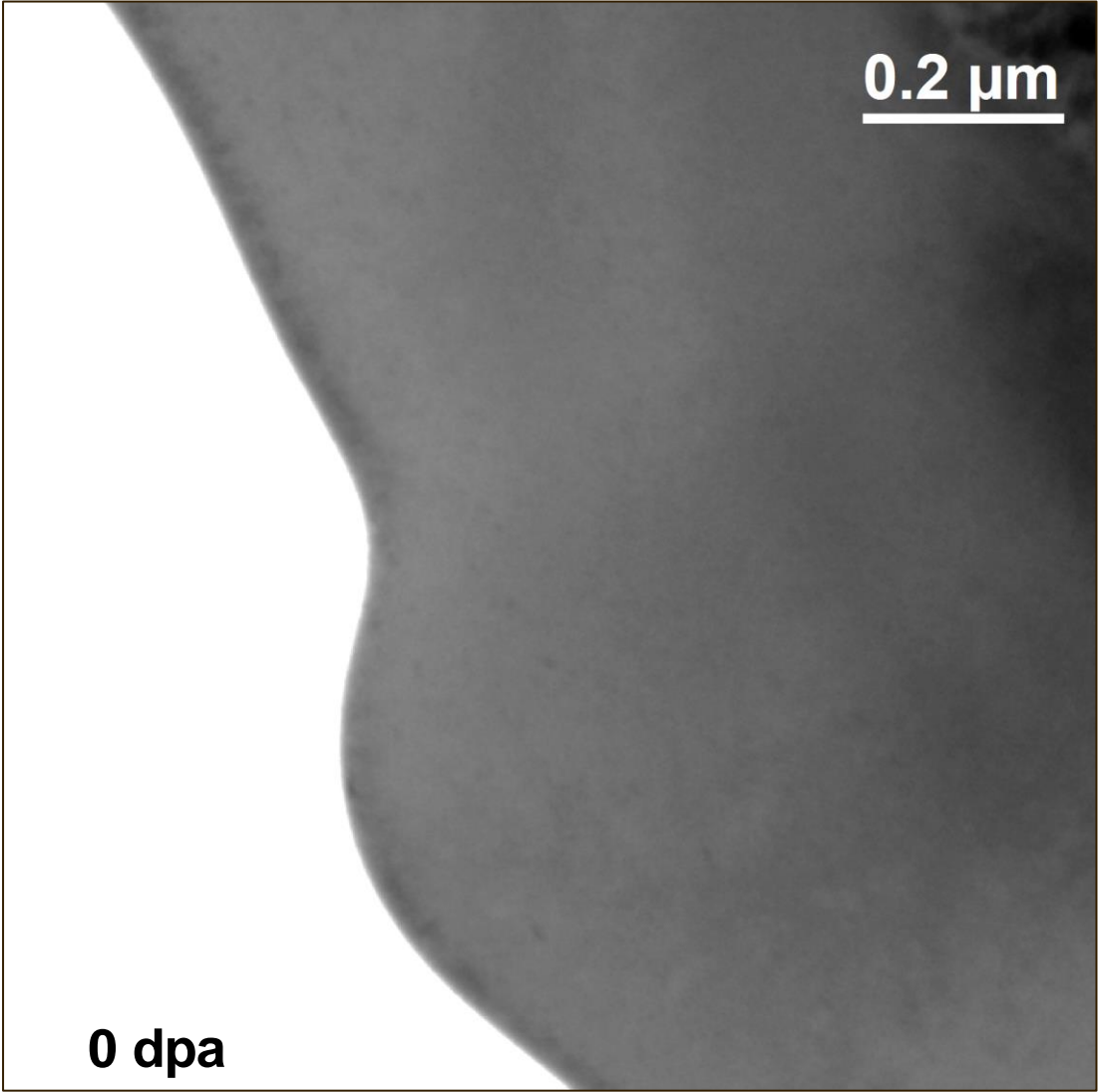
# ANL-IVEM Irradiations at 573 K: ~0.1 dpa



# ANL-IVEM Irradiations at 573 K: ~1.0 dpa



# ANL-IVEM Irradiations at 1073 K



# Summary of the Project

- A novel class of ternary silicides is emerging, thus extending the MAX phase concept beyond carbides and nitrides.
- Challenge is to **produce a pure ZIA phase**: as so it is for MAX phases.
- Refractory nanolayered intermetallic-ceramic compounds. Mechanical properties? Physical-chemical properties? And so on...
- Promising application in high-temperature irradiation environments. (although we tested only as-cast sample seemly off equilibrium/stabilization).



Materials at Extremes

Thank you!



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ZIA-PHASES

| [tunes@lanl.gov](mailto:tunes@lanl.gov) |

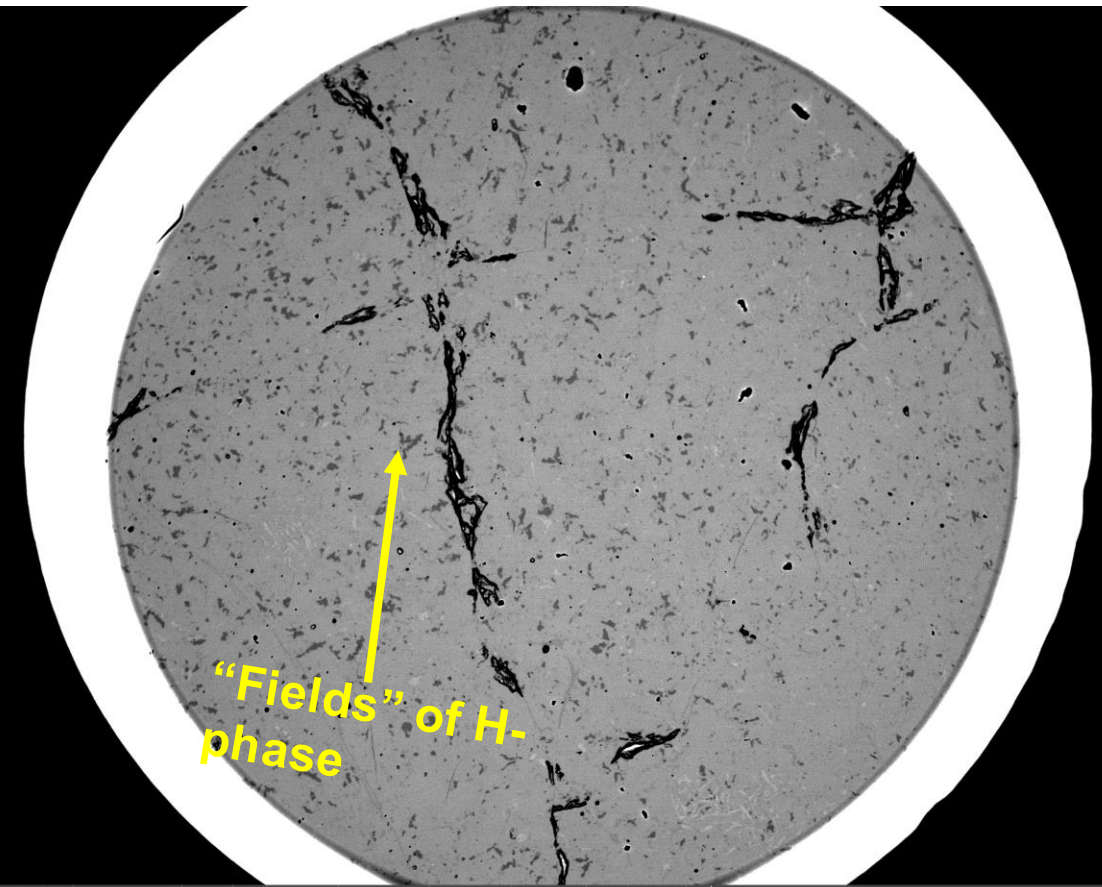
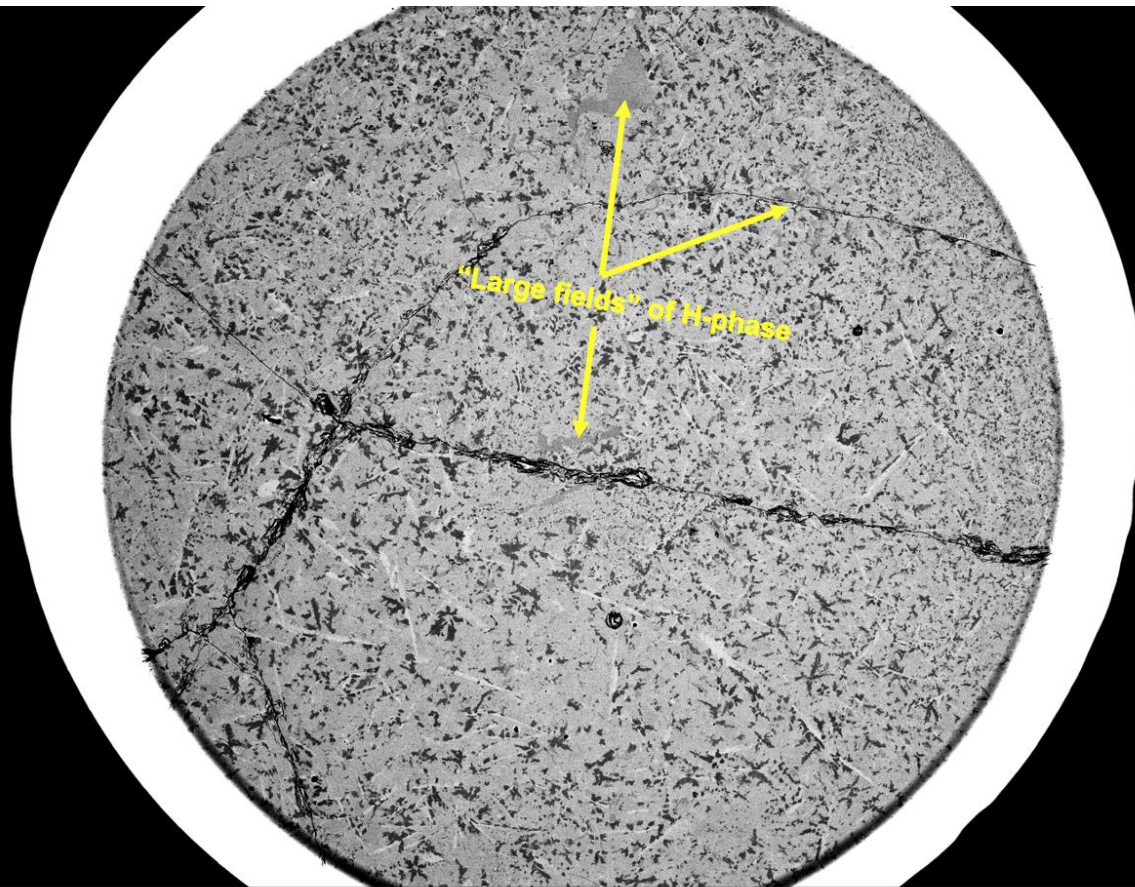
| [materialsatextremes.wordpress.com](http://materialsatextremes.wordpress.com) |



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# HT results



threshold use case det HV WD curr mag dwell  
0 eV Standard CBS 20.00 kV 9.9 mm 1.6 nA 75 x 10.00 μs

1 mm

threshold use case det HV WD curr mag dwell  
0 eV Standard CBS 20.00 kV 8.3 mm 3.2 nA 76 x 10.00 μs

1 mm  
HEAs WTaCrV

- 800°C for 120 hours: Gladyshevskii et al. 1962
- 800°C for 1000 hours: us, 2019.
- 1050°C for 336 hours: us, 2020.
- 1148°C for 336 hours: LANL, 2022.



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# XRD as-cast and HT

